# Rupestines F-M, New Guaipyridine Sesquiterpene Alkaloids from Artemisia rupestris 

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Eight new guaipyridine sesquiterpene alkaloids, rupestines $F-M(1-8)$ were isolated from the leaves of Artemisia rupestris and their structures were elucidated on the basis of 2D-NMR data. The absolute configurations of $1-8$ have been assigned by comparison of their experimental and calculated circular dichroism (CD) spectra.

Key words Artemisia rupestris; rupestine; guaipyridine sesquiterpene alkaloid; circular dichroism; absolute configuration

Artemisia rupestris L. (Compositae) is a well-known traditional Chinese medicinal plant in Xinjiang Province of China used for detoxification, antitumor, antibacterial, antivirus, and protecting liver. ${ }^{1,2)}$ It is also a well-known rich source of unique sesquiterpenes such as rupestonic acid showing anti-virus activity. ${ }^{3}$ ) Recently, five new sesquiterpene alkaloids, which have attracted great attention in the biogenetic and biological points of view, have been isolated from the flower of this plant. ${ }^{4,5)}$ In our further efforts to search for the structurally interesting metabolites from A. rupestris, eight new guaipyridine sesquiterpene alkaloids, rupestines $\mathrm{F}-\mathrm{M}$ $(\mathbf{1}-\mathbf{8})$ were isolated from the leaves. We applied density functional theory (DFT) calculations for determination of absolute configuration (AC) of $\mathbf{1 - 8}$, after conformational analysis of them. In this paper, we described the isolation and structure elucidation including ACs of $\mathbf{1 - 8}$ with pyridine cycloheptene ring system (Chart 1).

The alkaloid fraction prepared from acid-base solvent partitions of the methanol extract from the leaves of $A$. rupestris was separated using silica gel column, Sephadex LH-20 column, amino silica gel column, and preparative HPLC to give eight sesquiterpene alkaloids, rupestines $\mathrm{F}-\mathrm{M}(\mathbf{1}-\mathbf{8})$.

Rupestine F (1) had molecular formula of $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{2}$ determined by high resolution-electrospray ionization-mass spectrometry (HR-ESI-MS) [ $\left.\mathrm{m} / \mathrm{z} 260.1671(\mathrm{M}+\mathrm{H})^{+}\right]$. The UV absorption maxima at 270 nm was characteristic of a typical alkyl-substituted pyridine. ${ }^{6)}$ IR absorptions suggested the presence of a carbonyl $\left(1723 \mathrm{~cm}^{-1}\right)$ and a pyridine unit (1591, $1461 \mathrm{~cm}^{-1}$ ). The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 1) revealed a typical AB pattern for the protons at $\delta_{\mathrm{H}} 6.93(\mathrm{~d}, 7.6)$ and $\delta_{\mathrm{H}} 7.31$ (d, 7.6), one methyl group attached to the pyridine nucleus resonated at $\delta_{\mathrm{H}} 2.48,{ }^{7)}$ one methoxy at $\delta_{\mathrm{H}} 3.76$, and two singlet signals at $\delta_{\mathrm{H}} 5.61$ and 6.19 belonging to exo-methylene protons. ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Table 2) revealed 16 carbon signals due to one carbonyl, four $s p^{2}$ quaternary carbons, two $s p^{2}$ methines, one $s p^{2}$ methylene, two $s p^{3}$ methines, three $s p^{3}$ methylenes, and three methyls.

Partial structures of C-5 to C-9 and C-16 were deduced from analysis of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ correlation spectroscopy (COSY) and heteronuclear multiple bond connectivity (HMBC) spectra (Fig. 1). The HMBC correlations of $\mathrm{H}_{3}-15$ to $\mathrm{C}-2$ and $\mathrm{C}-3$, $\mathrm{H}-3$ to $\mathrm{C}-11$, and $\mathrm{H}-4$ to $\mathrm{C}-10$ showed that the existence of a tri-substituted pyridine ring (Fig. 1). The HMBC cross-peaks of $\mathrm{H}_{3}-16$ and $\mathrm{H}_{2}-9$ to $\mathrm{C}-11$, and $\mathrm{H}_{2}-9$ to $\mathrm{C}-10$ confirmed that a


1


5


2


6


3



8

Chart 1

Table 1. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ Data $\left[\delta_{\mathrm{H}}(J, \mathrm{~Hz})\right]$ of Rupestines $\mathrm{F}-\mathrm{M}(\mathbf{1}-\mathbf{8})$ in $\mathrm{CDCl}_{3}$

|  | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | 6.93 (d, 7.6) | 6.93 (d, 7.9) | 6.99 (d, 7.9) | 6.97 (d, 8.0) | 6.95 (d, 7.6) | 7.02 (d, 8.0) | 6.97 (d, 8.0) | 6.97 (d, 8.0) |
| 4 | 7.31 (d, 7.6) | 7.30 (d, 7.9) | 7.43 (d, 7.9) | 7.36 (d, 8.0) | 7.34 (d, 7.6) | 7.41 (d, 8.0) | 7.35 (d, 8.0) | 7.37 (d, 8.0) |
| 5 | 3.05 (m) | 2.99 (m) | 2.98 (m) | 3.01 (m) | 3.07 (m) | 3.02 (m) | 2.96 (m) | 2.96 (m) |
| $6 \alpha$ | $1.82(2 \mathrm{H}, \mathrm{m})$ | 2.01 (m) | 1.91 (m) | 1.79 (2H, m) | 1.82 (m) | 1.97 (m) | 1.67 (2H, m) | 1.86 (m) |
| $6 \beta$ |  | 1.76 (m) | 1.25 (m) |  | 1.72 (m) | 1.31 (m) |  | 1.34 (m) |
| $7 \alpha$ | 1.83 (2H, m) | 1.95 (m) | 1.86 (m) | 1.83 (m) | 1.83 (2H, m) | 1.97 (m) | 1.95 (m) | 1.87 (m) |
| $7 \beta$ |  | 2.12 (m) | 2.05 (m) | 2.02 (m) |  | 2.03 (m) | 2.05 (m) | 2.18 (m) |
| 8 | 2.83 (m) | 2.64 (m) | 2.58 (m) | 2.80 (m) | 2.42 (m) | 2.58 (m) | 4.13 (m) | 3.78 (m) |
| $9 \alpha$ | $\begin{aligned} & 3.24 \text { (dd, } 14.4, \\ & 10.0) \end{aligned}$ | $\begin{aligned} & 3.36(\mathrm{dd}, 14.4, \\ & 10.0) \end{aligned}$ | $\begin{aligned} & 3.28(\mathrm{dd}, 14.0, \\ & 10.2) \end{aligned}$ | $\begin{aligned} & 3.39 \text { (dd, 14.4, } \\ & 8.0) \end{aligned}$ | $\begin{aligned} & 3.27 \text { (dd, } 14.4 \text {, } \\ & 8.0 \text { ) } \end{aligned}$ | $\begin{aligned} & 3.35(\mathrm{dd}, 14.0, \\ & 10.8) \end{aligned}$ | 3.33 (s) | $\begin{aligned} & 3.34(\mathrm{dd}, 13.3, \\ & 10.3) \end{aligned}$ |
| $9 \beta$ | $\begin{aligned} & 3.15 \text { (dd, 14.4, } \\ & 2.8) \end{aligned}$ | 3.29 (d, 14.4) | 3.17 (d, 14.0) | $\begin{aligned} & 3.28 \text { (dd, 14.4, } \\ & 3.2) \end{aligned}$ | 3.24 (m) | 3.12 (d, 14.0) | 3.33 (d, 11.0) | 3.22 (d, 13.3) |
| 13a |  |  | $\begin{aligned} & 2.81 \text { (ddd, 17.6, } \\ & 6.5,4.0) \end{aligned}$ | $\begin{aligned} & 2.89 \text { (dt, 17.2, } \\ & 5.2) \end{aligned}$ | 4.14 (2H, s) | 4.39 (2H, s) |  |  |
| 13b |  |  | $\begin{aligned} & 2.74 \text { (ddd, 17.6, } \\ & 6.5,4.0) \end{aligned}$ | $\begin{aligned} & 2.66(\mathrm{dt}, 17.2 \text {, } \\ & 5.2) \end{aligned}$ |  |  |  |  |
| 14a | 5.61 (s) |  | $\begin{aligned} & 3.88 \text { (ddd, 11.0, } \\ & 6.5,4.0) \end{aligned}$ | 3.82 (2H, t, 5.2) | 5.05 (s) |  |  |  |
| 14b | 6.19 (s) |  | $\begin{aligned} & 3.86 \text { (ddd, 11.0, } \\ & 6.5,4.0) \end{aligned}$ |  | 4.94 (s) |  |  |  |
| 15 | 2.48 (s) | 2.47 (s) | 2.49 (s) | 2.46 (s) | 2.50 (s) | 2.51 (s) | 2.49 (s) | 2.50 (s) |
| 16 | 1.33 (d, 7.2) | 1.33 (d, 7.2) | 1.35 (d, 7.6) | 1.36 (d, 7.2) | 1.37 (d, 7.2) | 1.37 (d, 7.2) | 1.35 (d, 7.2) | 1.34 (d, 7.1) |
| OMe | 3.76 (s) | 3.65 (s) |  |  |  |  |  |  |

Table 2. ${ }^{13} \mathrm{C}-\mathrm{NMR}$ Data $\left(\delta_{\mathrm{C}}\right)$ of $\mathbf{1}-\mathbf{8}$ in $\mathrm{CDCl}_{3}$

| No. | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | 154.6 | 154.9 | 154.5 | 154.6 | 154.5 | 154.2 | 154.6 | 154.7 |
| 3 | 121.1 | 121.3 | 121.5 | 121.5 | 121.1 | 121.5 | 121.5 | 121.3 |
| 4 | 136.7 | 136.2 | 133.1 | 135.2 | 135.4 | 132.6 | 134.2 | 132.9 |
| 5 | 37.9 | 37.7 | 35 | 36.7 | 36.9 | 34.8 | 36.4 | 35.3 |
| 6 | 34 | 32.3 | 34.9 | 32.4 | 33.2 | 34.8 | 30.4 | 33.5 |
| 7 | 31.7 | 29.8 | 32.6 | 28.6 | 31.6 | 33.3 | 36.4 | 39.2 |
| 8 | 38 | 42.2 | 49.4 | 48.7 | 40 | 45.2 | 66.9 | 68.6 |
| 9 | 43.7 | 40.8 | 38.8 | 38.4 | 42.4 | 39.5 | 46.1 | 48.4 |
| 10 | 158.5 | 157.5 | 158.9 | 157.2 | 158.7 | 159.3 | 156.2 | 156.6 |
| 11 | 137.6 | 137.8 | 138.1 | 138 | 137.5 | 137.7 | 138.5 | 138.5 |
| 12 | 146.2 | 175.9 | 214 | 213.9 | 153.8 | 208.9 |  |  |
| 13 | 167.5 |  | 43.2 | 43.8 | 65.1 | 66.8 |  |  |
| 14 | 123.2 |  | 58.2 | 58.3 | 110.1 |  |  |  |
| 15 | 23.9 | 24 | 23.6 | 23.6 | 23.8 | 23.6 | 23.8 | 23.9 |
| 16 | 18.3 | 18.9 | 20.5 | 19.1 | 18.8 | 20.4 | 19.7 | 20.1 |
| OMe | 51.9 | 51.7 |  |  |  |  |  |  |

cycloheptane fused to the pyridine ring at $\mathrm{C}-10$ and $\mathrm{C}-11$. The correlations of $\mathrm{H}_{3}-\mathrm{OMe}$ to $\mathrm{C}-13$, and $\mathrm{H}_{2}-14$ to $\mathrm{C}-8, \mathrm{C}-12$, and $\mathrm{C}-13$ unequivocally established that the position of an isopropenoic acid methyl ester group was allowed to C-8.

The relative configuration of $\mathbf{1}$ was established by nuclear Overhauser effect spectroscopy (NOESY) correlations as shown in computer-generated 3D drawing (Fig. 2). The ${ }^{3} J$ proton coupling constant ( ${ }^{3} J_{\mathrm{H}-9 \alpha / \mathrm{H}-8}=10.0 \mathrm{~Hz}$ ) as well as NOESY correlations of $\mathrm{H}-9 \alpha / \mathrm{H}_{3}-16$ and $\mathrm{H}-9 \beta / \mathrm{H}-8$ indicated that each of $\mathrm{C}-12$ and $\mathrm{C}-16$ adopted an $\alpha$-configuration as shown in Fig. 2. Thus, the structure of $\mathbf{1}$ was established to be shown.

Rupestine G (2) gave a molecular formula of $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{2}$ as determined by the HR-ESI-MS at $m / z 234.1509(\mathrm{M}+\mathrm{H})^{+}$. IR absorptions implied the presence of carbonyl $\left(1736 \mathrm{~cm}^{-1}\right)$ functionality. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR data (Tables 1, 2) revealed 14


1


2

## - ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY <br> $\rightarrow \mathrm{HMBC}$

Fig. 1. Selected ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and HMBC Correlations for Rupestines F-G (1-2)


Fig. 2. Selected NOESY Correlations for Rupestines F (1) and H (3)
carbon signals due to three $s p^{2}$ quaternary carbons, two $s p^{2}$ methines, two $s p^{3}$ methines, three $s p^{3}$ methylenes, and three methyls (including one methoxy). Among them, three $s p^{2}$ quaternary carbons ( $\delta_{\mathrm{C}} 154.9,157.5,137.8$ ) and two $s p^{2}$ methines ( $\delta_{\mathrm{C}} 121.3,136.2$ ) revealed the presence of a trisubstituted pyridine ring. The ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of 2 were quite similar to those of $\mathbf{1}$, which suggested that $\mathbf{2}$ also possessed a guaipyridine sesquiterpene skeleton.

The connectivities of C-16 to C-9 were confirmed by ${ }^{1} \mathrm{H}{ }^{1} \mathrm{H}$ COSY spectrum (Fig. 1). The HMBC correlations of $\mathrm{H}-9$ to $\mathrm{C}-10$ and $\mathrm{C}-11$, and $\mathrm{H}_{3}-16$ to $\mathrm{C}-11$ gave rise to the connectivity of cycloheptene and pyridine rings through $\mathrm{C}-10$ and $\mathrm{C}-11$. The methoxy carbonyl side chain was deduced from the HMBC correlations of $\mathrm{H}-9$ and $\mathrm{H}_{3}-\mathrm{OMe}$ to $\mathrm{C}-12$. Based on the NOESY correlations between $\mathrm{H}-9 \alpha$ and $\mathrm{H}_{3}-16$, and the ${ }^{3} J$ proton coupling constant ( ${ }^{3} J_{\mathrm{H}-9 \alpha / \mathrm{H}-8}=10.0 \mathrm{~Hz}$ ), the relative configuration of $\mathrm{C}-5$ and $\mathrm{C}-8$ of $\mathbf{2}$ was concluded to be the same as $\mathbf{1}$.

Rupestine H (3) had the composition of $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{2}$ by HR-ESI-MS [m/z $248.1665(\mathrm{M}+\mathrm{H})^{+}$]. The ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data of $\mathbf{3}$ were analogous to those of $\mathbf{1}$ except for the substituent at $\mathrm{C}-8$, and the ${ }^{13} \mathrm{C}$ resonances of one carbonyl ( $\mathrm{C}-12, \delta_{\mathrm{C}} 214.0$ ) and two $s p^{3}$ methylenes ( $\mathrm{C}-13, \delta_{\mathrm{C}} 43.2 ; \mathrm{C}-14, \delta_{\mathrm{C}} 58.2$ ) were additionally observed in 3. Based on the 2D-NMR data, the side chain of C-8 was deduced to be a 3-hydroxypropan-1-one moiety. The relative configuration of $\mathbf{3}$ was determined by a key rotating frame Overhauser enhancement spectroscopy (ROESY) (Fig. 2) correlation of $\mathrm{H}-5 \alpha / \mathrm{H}-9 \alpha$ and the ${ }^{3} J$ proton coupling constant ( ${ }^{3} J_{\mathrm{H}-9 \alpha / \mathrm{H}-8}=10.2 \mathrm{~Hz}$ ) which revealed that $\mathrm{C}-16$ was $\beta$-oriented and $\mathrm{C}-12$ was $\alpha$-oriented.

Rupestine I (4) possessed a molecular formula, $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{2}$ determined by HR-ESI-MS $\left[\mathrm{m} / \mathrm{z} 248.1668(\mathrm{M}+\mathrm{H})^{+}\right] .{ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1,2 ) of 4 were similar to those of $\mathbf{3}$, implying the isomeric nature for $\mathbf{3}$ and 4 . 4 was assumed to be an epimer of $\mathbf{3}$, which was confirmed by the NOESY correlation between $\mathrm{H}-9 \alpha$ and $\mathrm{H}_{3}-16$.

Rupestine J (5) had a molecular formula, $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}$ determined by the HR-ESI-MS $\left[m / z 232.1724(\mathrm{M}+\mathrm{H})^{+}\right] .{ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1,2) of 5 revealed an exo-methylene group ( $\delta_{\mathrm{H}} 5.05,4.94, \delta_{\mathrm{C}} 110.1$ ). Based on the 2D-NMR data, the structure of $\mathbf{5}$ was assigned to be similar to $\mathbf{1}$. The major difference was the presense of a hydroxy methyl ( $\delta_{\mathrm{C}} 65.1$ ) at $\mathrm{C}-12$ in $\mathbf{5}$ instead of a methoxycarbonyl group in 1. Relative configuration of $\mathrm{C}-12$ and $\mathrm{C}-16$ was confirmed to be cis, since the NOESY correlation of $\mathrm{H}-9 \alpha / \mathrm{H}_{3}-16$ was observed.

Rupestine K (6) was assigned the molecular formula $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{2}$ through an analysis of its HR-ESI-MS [ $\mathrm{m} / \mathrm{z}$ $\left.234.1517(\mathrm{M}+\mathrm{H})^{+}\right]$. The NMR spectra (Tables 1, 2) of 6 closely matched with those of $\mathbf{3}$. A significant difference was the
presence of a hydroxy methyl at $\delta_{\mathrm{C}} 66.8$ connected to $\mathrm{C}-12$ in 6 instead of a hydroxy ethyl in 3. NOESY correlation of H-5 $\alpha /$ $\mathrm{H}-9 \alpha$ and the ${ }^{3} J$ proton coupling constant $\left({ }^{3} J_{\mathrm{H}-9 \alpha / \mathrm{H}-8}=10.8 \mathrm{~Hz}\right)$ suggested that the relative configurations of C-5 and C-8 in 6 were identical to those of $\mathbf{3}$.

The molecular formula of rupestines $L$ (7) and $M(8)$ obtained as a pair of epimer, was determined to be $\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{NO}$ by analysis of the HR-ESI-MS and the NMR spectra (Tables 1, 2). Comparison of the NMR data of $\mathbf{1}$ and $\mathbf{8}$ indicated that the isopropenoic acid methyl ester group at C-8 in $\mathbf{1}$ was replaced by a hydroxy group in 8 , which was confirmed by COSY and HMBC experiments. Configuration of C-16 in $\mathbf{8}$ was found to be $\beta$-oriented on the basis of a ROESY cross-peak of $\mathrm{H}-5 \alpha / \mathrm{H}-$ $9 \alpha$ and the ${ }^{3} J$ proton coupling constant $\left({ }^{3} J_{\mathrm{H}-9 \alpha / \mathrm{H}-8}=10.3 \mathrm{~Hz}\right)$. On the other hand, $\mathrm{C}-16$ in 7 was assigned to be $\alpha$-oriented, because ROESY correlations of $\mathrm{H}-9 \alpha / \mathrm{H}_{3}-16$ and $\mathrm{H}-9 \beta / \mathrm{H}-8$ were observed.
The AC of $\mathbf{1 - 8}$ was assigned by using circular dichroism (CD) spectroscopy and time dependent density functional theory (TDDFT) calculations at B3LYP/TZVPP level. The CD spectra of $\mathbf{2 - 8}$ (see Experimental) showed a similar CD pattern, i.e. a negative Cotton effect (CE) near 270 nm and a positive CE in the $210-225 \mathrm{~nm}$ region, whereas some compounds exhibited a positive CE around 285 nm which may be attributed to the carbonyl group. The CE near 270 nm can be attributed to the pyridine ring, and thus the sign of this CE may be governed by the AC of $\mathrm{C}-5$, by the preferred conformation, or by both the AC of C-5 and the preferred conformation. In contrast, the CD spectrum of $\mathbf{1}$ showed a different CD pattern, a negative CE at 213 nm and weak positive CEs above 220 nm . The difference may probably be caused by the additional $\alpha, \beta$ unsaturated ketone chromophore in $\mathbf{1}$.

To study the effect of conformation to the CE sign at 270 nm , for possible conformations of model compounds 2-methylpyridinecycloheptene (9), 7 and $\mathbf{8}$, their corresponding CD spectra were calculated (Figs. 3, 4). The calculation results are summarized as follow, (1) B3LYP/TZVPP calculations overestimated the excitation energy of pyridine's first absorption band (calc. ca. 250 nm vs. expt. ca. 270 nm ), thus, the calculated CE around 250 nm will be observed experimentally around $c a .270 \mathrm{~nm}$; (2) if the plane of C-6, C-7, and C-8 was above the pyridine ring, a positive CE around 215 nm and a negative CE around 250 nm are observed regardless of the chirality of C-5. Thus, using the CE sign around 270 nm , the absolute structure taken by the molecule can be deduced and the AC of C-5 and C-8 may be assigned accordingly.

Based on this conclusion, the AC of $\mathbf{2 - 8}$ was then assigned. According to the larger coupling between $\mathrm{H}-9 \alpha$ and


7


9a


8


9b

Fig. 3. Calculated Conformations of Model Compounds


Fig. 4. Calculated CD Spectra of Model Compounds Together with the Experimental CD Spectra of $\mathbf{7}$ and $\mathbf{8}$


Fig. 5. Experimental CD Spectrum of $\mathbf{3}$ and Calculated CD Spectrum of the $(5 S, 8 S)$ of $\mathbf{3}$

H-8 (Tables 1, 2) and conformational analysis by molecular mechanics calculation, $\mathbf{2 - 8}$ existed mainly in the chair conformation with the substituent at $\mathrm{C}-8$ in the equatorial position. Experimental CD spectra of $\mathbf{2 - 8}$ implied the position of C-6, C-7, and C-8 to be above the pyridine ring, thus the configuration of $\mathrm{C}-8$ should be $S$ for $\mathbf{2}-\mathbf{8}$.

It has been mentioned that some compounds with a ketone group at C-12 exhibited a positive CE around $285 \mathrm{~nm}(\mathbf{2 - 4}$, 6). Among these compounds, 3 was chosen as a representative and its AC was assigned by comparing the calculated CD spectra to the experimental one (Fig. 5).

The AC of 1, which has a different CD pattern, was assigned by comparing the calculated CD spectra to the ex-


Fig. 6. Experimental CD Spectrum of $\mathbf{1}$ and Calculated CD Spectrum of the $(5 S, 8 S)$ of $\mathbf{1}$
perimental one (Fig. 6). The calculated CD spectrum of the $(5 S, 8 S)$ isomer is similar to the experimental one, thus the AC of $\mathbf{1}$ was assigned as $(5 S, 8 S)$.

## Experimental

General Experimental Procedures The UV spectra were obtained with a Ultrospec 2100 pro spectrophotometer. Optical rotations were measured with a JASCO DIP-1000 automatic digital polarimeter, and CD spectra were measured on a JASCO J-820 spectropolarimeter. IR spectra were recorded on a JASCO FT/IR-4100 spectrophotometer. High-resolution ESI-MS were obtained on a LTQ Orbitrap XL (Thermo Scientific). ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR spectra were recorded on a

JEOL ECA600 spectrometer and a Bruker AV 400 spectrometer, and chemical shifts are referenced to the residual solvent peaks ( $\delta_{\mathrm{H}} 3.31$ and $\delta_{\mathrm{C}} 49.0$ for methanol $-d_{4}$ and $\delta_{\mathrm{H}} 7.26$ and $\delta_{\mathrm{C}}$ 77.0 for $\mathrm{CDCl}_{3}$ ). Standard pulse sequences were employed for the 2D-NMR experiments.

Material The plant material of $A$. rupestris $L$. were collected from Hami City, Xinjiang Province, P.R. China, in June 2008, and was authenticated by Prof. Shi-Ming Duan (Xinjiang Institute of Ecology and Geography, Chinese Academy of Sciences). A voucher specimen has been deposited with Xinjiang Technical Institute of Physics and Chemistry, Chinese Academy of Sciences, P. R. China.

Extraction and Isolation The air dried and milled leaves of A. rupestris L. ( 10 kg ) were extracted by maceration with $\mathrm{MeOH}(4 \times 60 \mathrm{~L}$, each time 2 d$)$ at r.t. After filtration and evaporation of solvent under reduced pressure, the combined methanol extract ( 2.29 kg ) was partitioned between $3 \%$ tartaric acid and $\mathrm{CHCl}_{3}$ to obtain a water soluble fraction. The water soluble fraction was adjusted to pH 10 by saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$, then partitioned with $\mathrm{CHCl}_{3}$ and $n-\mathrm{BuOH}$, successively, to give $\mathrm{CHCl}_{3}$ and $n$ - BuOH fractions

The $\mathrm{CHCl}_{3}$ extract was chromatographed over $\mathrm{CC}\left(\mathrm{SiO}_{2}\right.$; hexane/EtOAc, 8:1 $\rightarrow 2: 1$, and $\mathrm{CHCl}_{3} / \mathrm{MeOH} 30: 1 \rightarrow 0: 1$ ) to afford 32 fractions. Fr. 4 was chromatographed over Sephadex LH-20 eluting with MeOH , then subjected to purification by preparative $\mathrm{HPLC}\left(\mathrm{C}_{18}, \mathrm{MeOH} / 0.1 \% \mathrm{HCOOH}\right.$ aq., $2: 3$ ) to afford rupestine $\mathrm{F}(\mathbf{1}, 4.7 \mathrm{mg})$. Fr. 9 was purified by preparative HPLC ( $\mathrm{C}_{18}, \mathrm{MeOH} / 0.1 \% \mathrm{HCOOH}$ aq., $3: 7$ ) to yield rupestine G ( $2,0.5 \mathrm{mg}$ ). Fr. 15 was subjected to Sephadex LH-20 column $(\mathrm{MeOH})$ to give 4 fractions, then the second fraction was subjected over a $\mathrm{SiO}_{2}$ column (hexane/EtOAc 1:1, and then $\mathrm{CHCl}_{3} / \mathrm{MeOH} 20: 1 \rightarrow 1: 1$ ) to afford 8 subfractions.

Sub. Fr. 7 was purified with $\mathrm{C}_{18}$ column ( $\mathrm{MeOH} / 0.1 \%$ HCOOH aq., $35: 65$ ) followed by preparative HPLC $\left(\mathrm{C}_{18}\right.$, $\mathrm{MeOH} / 0.1 \% \mathrm{HCOOH}$ aq., $18: 82$ ) to give rupestines H (3, $0.8 \mathrm{mg})$ and $\mathrm{I}(4,1.9 \mathrm{mg})$. Sub. Fr. 4 was treated on amino silica gel (hexane/EtOAc, 1:1) and then purified by preparative HPLC ( $\mathrm{C}_{18}, \mathrm{MeOH} / 0.1 \% \mathrm{HCOOH}$ aq., $22: 78$ and $\mathrm{MeOH} / 0.1 \%$ HCOOH aq., $14: 86$ ) to afford rupestines $\mathrm{J}(5,0.7 \mathrm{mg})$, K (6, 0.9 mg ), L ( $7,0.8 \mathrm{mg}$ ), and M ( $8,0.5 \mathrm{mg}$ ).

The $n-\mathrm{BuOH}$ fraction was separated over Sephadex LH-20 column chromatography $(\mathrm{MeOH})$ to afford 4 fractions. Fr. 3 was further purified by an ODS column $(40 \% \mathrm{MeOH})$ to give a known alkaloid, rupestine ${ }^{4}$ ( 39.3 mg ).

Rupestine F (1): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-62 \quad(c=0.2$, $\mathrm{MeOH})$; UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 270$ (3595) nm; CD $\lambda_{\text {max }}$ $(\mathrm{MeOH}) \mathrm{nm}(\Delta \varepsilon): 271(+1.3), 246(-0.5), 213(-8.1)$; IR $v_{\max }$ $\left(\mathrm{CCl}_{4}\right) \mathrm{cm}^{-1}: 3057,2952,2927,2852,1723,1626,1591,1573$, 1461; ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS m/z 260 $[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS m/z $260.1671[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{NO}_{2}, 260.1651$ ).

Rupestine $G$ (2): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-16 \quad(c=0.03$, $\mathrm{MeOH}) ; \mathrm{UV} \lambda_{\max }(\mathrm{MeOH}) \mathrm{nm}$ ( $\varepsilon: 270$ (3760), 213 (5644); CD $\lambda_{\max }(\mathrm{MeOH}) \mathrm{nm}\left(\Delta_{\varepsilon}\right): 285(0.06) 268(-1.2), 224(+0.4) ;$ IR $v_{\max }\left(\mathrm{CCl}_{4}\right) \mathrm{cm}^{-1}: 2960,2926,2855,1736,1593,1463 ;{ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS m/z $234[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS m/z $234.1509[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{NO}_{2}$, 234.1519).

Rupestine H (3): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}+18(c=0.1, \mathrm{MeOH})$; UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 270$ (3939), 215 (5733) nm; CD $\lambda_{\text {max }}$ $(\mathrm{MeOH}) \mathrm{nm}(\Delta \varepsilon): 284(+1.5), 268(-5.0), 216(+1.4) ;$ IR $v_{\max }$
$\left(\mathrm{CCl}_{4}\right) \mathrm{cm}^{-1}$ : 2959, 2925, 2854, 1707, 1463; ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS $m / z 248[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS $m / z$ $248.1665[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NO}_{2}, 248.1651$ ).

Rupestine I (4): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-12(c=0.1, \mathrm{MeOH})$; UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 270(3997) \mathrm{nm} ; \mathrm{CD} \lambda_{\text {max }}(\mathrm{MeOH})$ $\mathrm{nm}(\Delta \varepsilon): 285(+0.4), 267(-6.9), 218(-1.0)$; IR $v_{\max }\left(\mathrm{CCl}_{4}\right)$ $\mathrm{cm}^{-1}: 2958,2926,2855,1708,1951,1462 ;{ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS $m / z 248[\mathrm{M}+\mathrm{H}]^{+} ;$HR-ESI-MS $m / z$ $248.1668[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NO}_{2}, 248.1651$ ).

Rupestine J (5): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-17$ ( $c=0.1, \mathrm{MeOH}$ ); UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 271$ (3989), 215 (6545) nm; CD $\lambda_{\text {max }}$ $(\mathrm{MeOH}) \mathrm{nm}\left(\Delta_{\varepsilon}\right): 270(-5.1), 224(+0.1)$; IR $v_{\max }\left(\mathrm{CCl}_{4}\right) \mathrm{cm}^{-1}$ : 2951, 2926, 2855, 1590, 1463; ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS $m / z 232[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS $m / z 232.1724$ $[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\left.\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NO}, 232.1701\right)$.

Rupestine K (6): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}+20 \quad(c=0.03$, $\mathrm{MeOH})$; UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 271$ (3712), 215 (5483) nm; $\mathrm{CD} \lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\Delta \varepsilon): 282(+2.2), 267(-4.8), 217(+2.0)$; IR $v_{\text {max }}\left(\mathrm{CCl}_{4}\right) \mathrm{cm}^{-1}: 2958,2925,2855,1710,1591,1463 ;{ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS m/z $234[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS $m / z 234.1517[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{NO}_{2}$, 234.1494).

Rupestine L (7): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-40 \quad(c=0.03$, $\mathrm{MeOH})$; UV $\lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 270$ (4026), 215 (5630) nm; $\mathrm{CD} \lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}\left(\Delta_{\varepsilon}\right): 267(-1.3), 218(+0.9)$; IR $\left(\mathrm{CCl}_{4}\right)$ $v_{\max } 2960,2926,2856,1590$, and $1464 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS $m / z 192[\mathrm{M}+\mathrm{H}]^{+}$; HR-ESI-MS m/z $192.1384[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{NO}, 192.1388$ ).
Rupestine M (8): Light yellow oil; $[\alpha]_{\mathrm{D}}^{20}-38(c=0.05$, $\mathrm{MeOH}) ; \mathrm{UV} \lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\varepsilon): 270$ (3900), 215 (5052) nm; $\mathrm{CD} \lambda_{\text {max }}(\mathrm{MeOH}) \mathrm{nm}(\Delta \varepsilon): 267(-3.0), 215(+2.3)$; IR $\left(\mathrm{CCl}_{4}\right)$ $v_{\max }$ 2961, 2925, 2856, and $1462 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data (Tables 1, 2); ESI-MS m/z 192 [M+H] ${ }^{+}$; HR-ESI-MS m/z $192.1378[\mathrm{M}+\mathrm{H}]^{+}$(Calcd for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{NO}, 192.1388$ ).

Computational Details The CD calculations performed by Turbomole $6.3^{8)}$ using TD-DFT-B3LYP/TZVPP level of theory on RI-DFT-B3LYP/TZVPP optimized geometries. ${ }^{9-15)}$ The conformer used for CD calculation was the model obtained by using MC calculations (MMFF94 force field, ${ }^{16}$ ) Macromodel 9.1). ${ }^{17)}$ The CD spectra were simulated by overlapping Gaussian functions for each transition where the width of the band at $1 / \mathrm{e}$ height is fixed at 0.3 eV or 0.25 eV , and the resulting spectra were scaled to the experimental values.

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